



VK-701 LEAP5™ Update

Sulfuric Acid Roundtable 2013

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VK-701 LEAP5™

- Designed for high SO₃ environments
 - 4th bed of single absorption plants
 - 3rd bed of double absorption plants
- Introduced at Sulfur Conference in 2010
- Presentation at Sulfuric Acid Roundtable in 2011

How to reduce SO₂ emissions

- 1. Process changes
- 2. Catalyst – more liters/STPD – aka more horsepower
- 3. Optimized catalyst – VK69 & VK-701

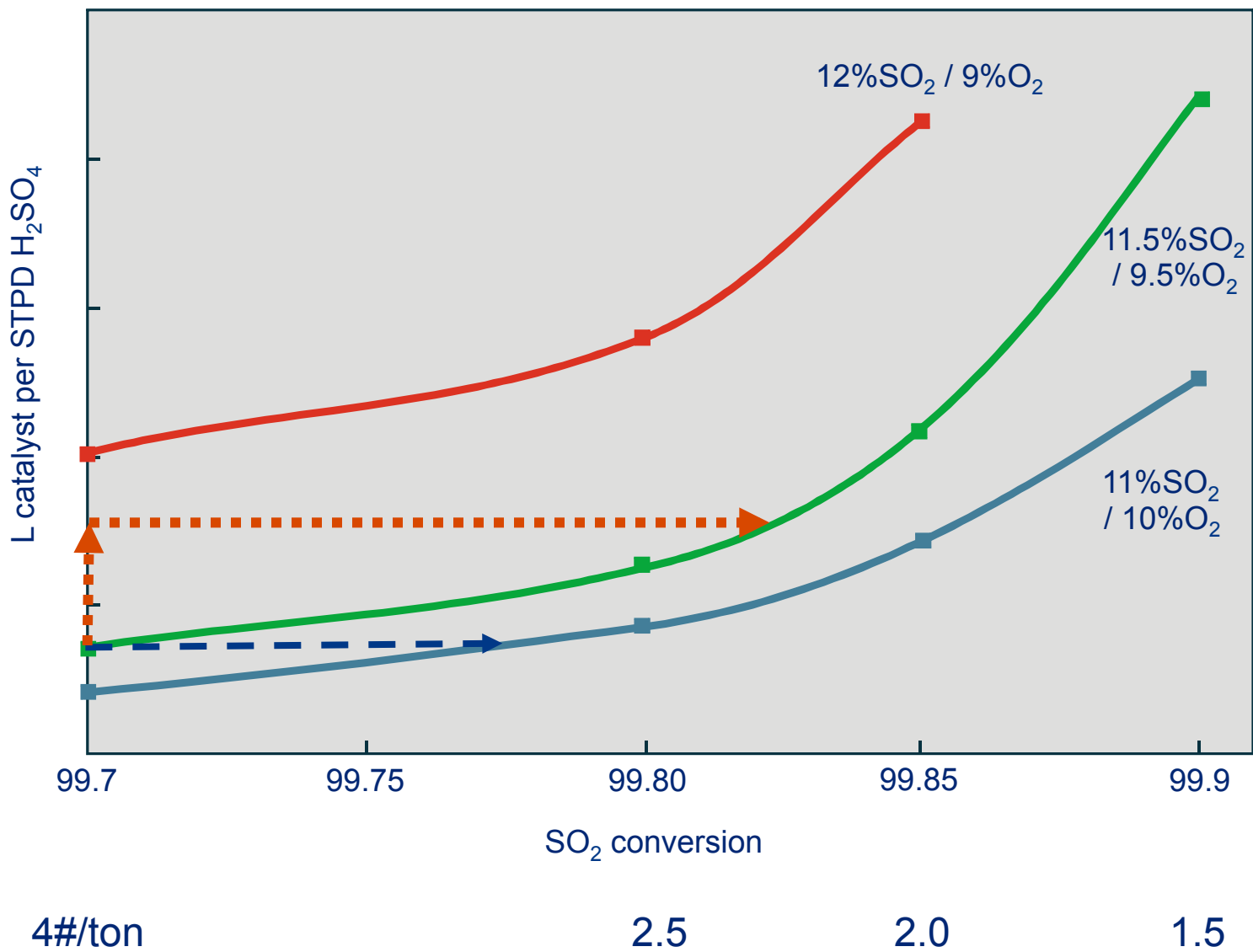
Process Options

- Reduce rate
- Reduce gas strength
- Single absorption convert to double absorption
- Install a scrubber - \$\$\$\$, annual O&M costs, possible waste disposal issues

Catalyst – increase liters/STPD

- Add more catalyst if there is room
- Add additional satellite bed & operate 2 existing beds in parallel
- Build a new converter & add more catalyst – space requirements and \$\$\$\$

Liters/STPD



Optimized Catalyst

- Catalyst optimized for specific gas compositions
 - Low SO_2
 - Low O_2
 - High SO_3

Catalyst Refresher



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Sulfuric Acid Catalyst Composition

- Vanadium
- Sodium
- Potassium
- Diatomaceous earth
- Binders

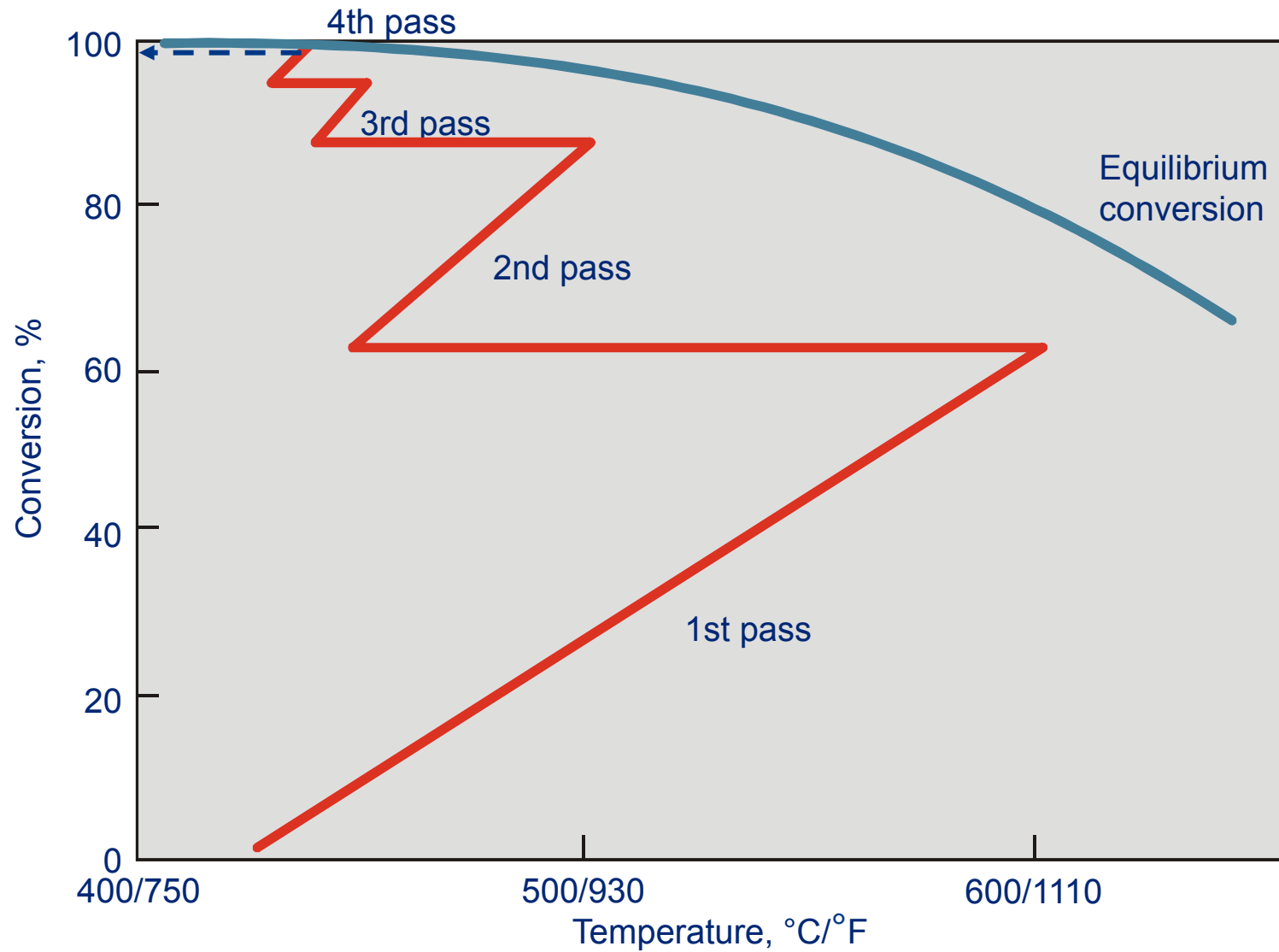
Supported Liquid Phase Catalyst

- Vanadium salts are liquid at operating temperature
- DE substrate provides physical structure
- Liquid melt coats the surface and pores of the DE

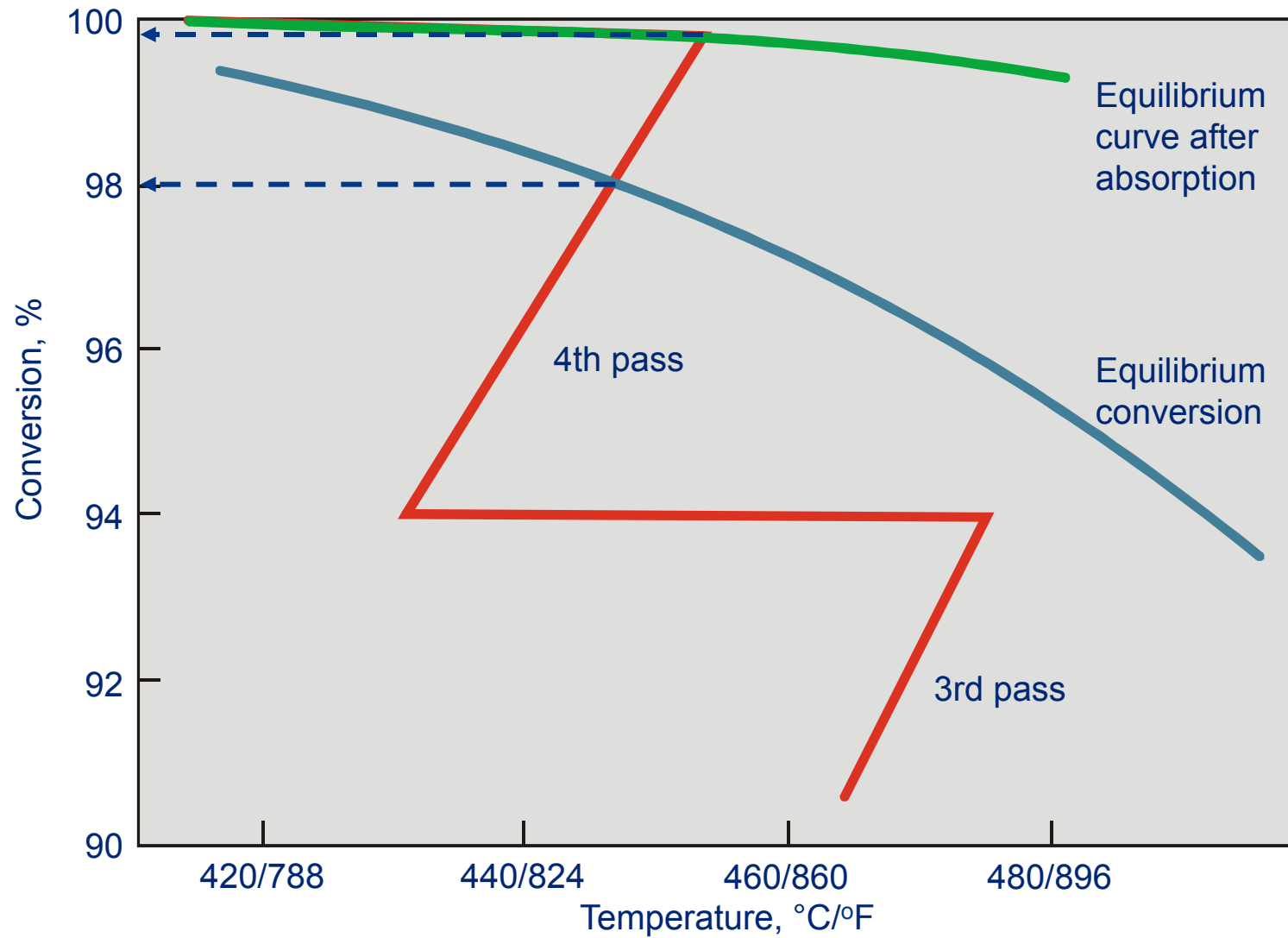
Conversion of SO₂ to SO₃

- Reversible reaction
- $\text{SO}_2 + \frac{1}{2}\text{O}_2 \rightleftharpoons \text{SO}_3$
 - Temperature
 - Pressure
 - Concentration
- Equilibrium – rate of forward reaction equals the rate of the reverse reaction
- Reaction rate slows down as approach equilibrium

Equilibrium and conversion



Equilibrium and Double Absorption



Optimized Catalyst



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Basis - 3x1 DA 11.5% SO₂ sulfur burner with 150 liters/STPD catalyst loading

Bed	SO ₂	O ₂	SO ₃	N ₂
1	11.5	9.45	0	79.05
2	4.77	6.22	7.14	81.87
3	1.78	4.78	10.31	83.13
4	1.00	4.90	0.01	94.09

Bed Specific catalyst

- The gas composition at the inlet to each catalyst bed varies significantly.
- For maximum conversion and minimum SO₂ emissions one size fits all catalyst is not sufficient
- In the future, plants may utilize a different catalyst in each bed to maximize conversion and reduce emissions

VK69

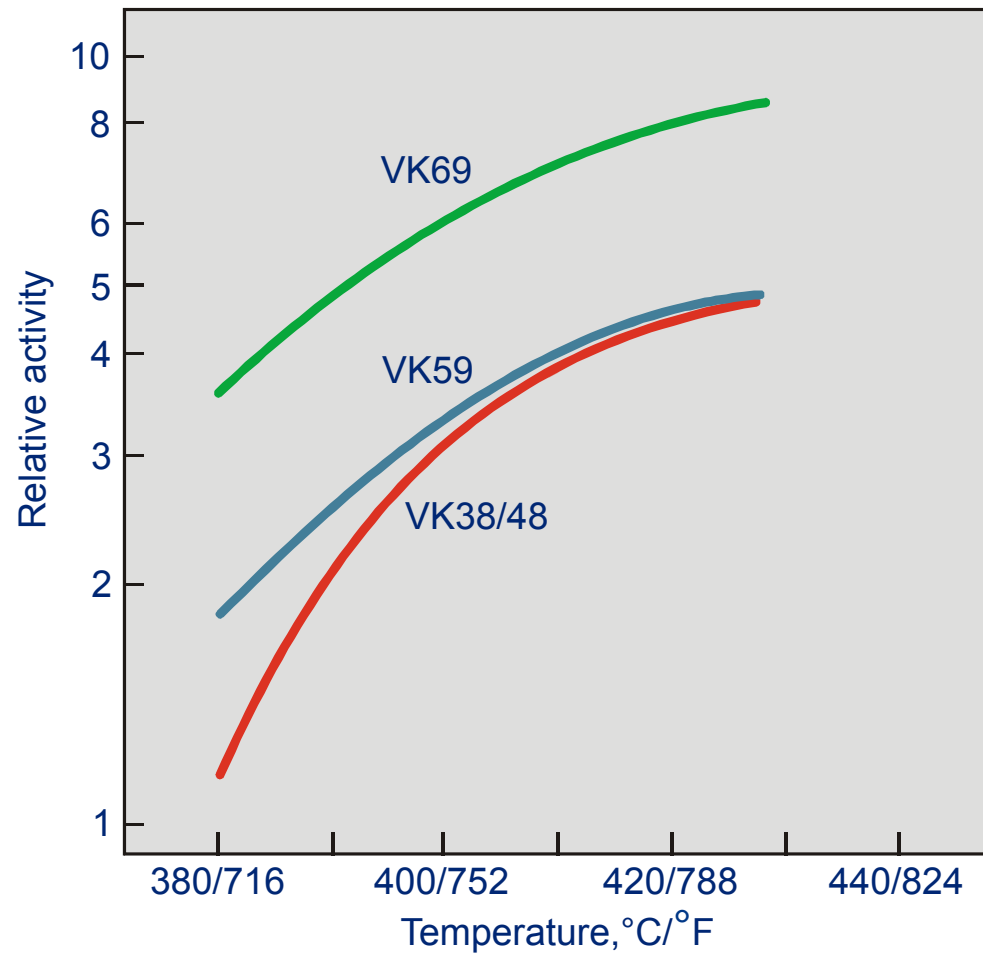


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VK69 cesium promoted catalyst

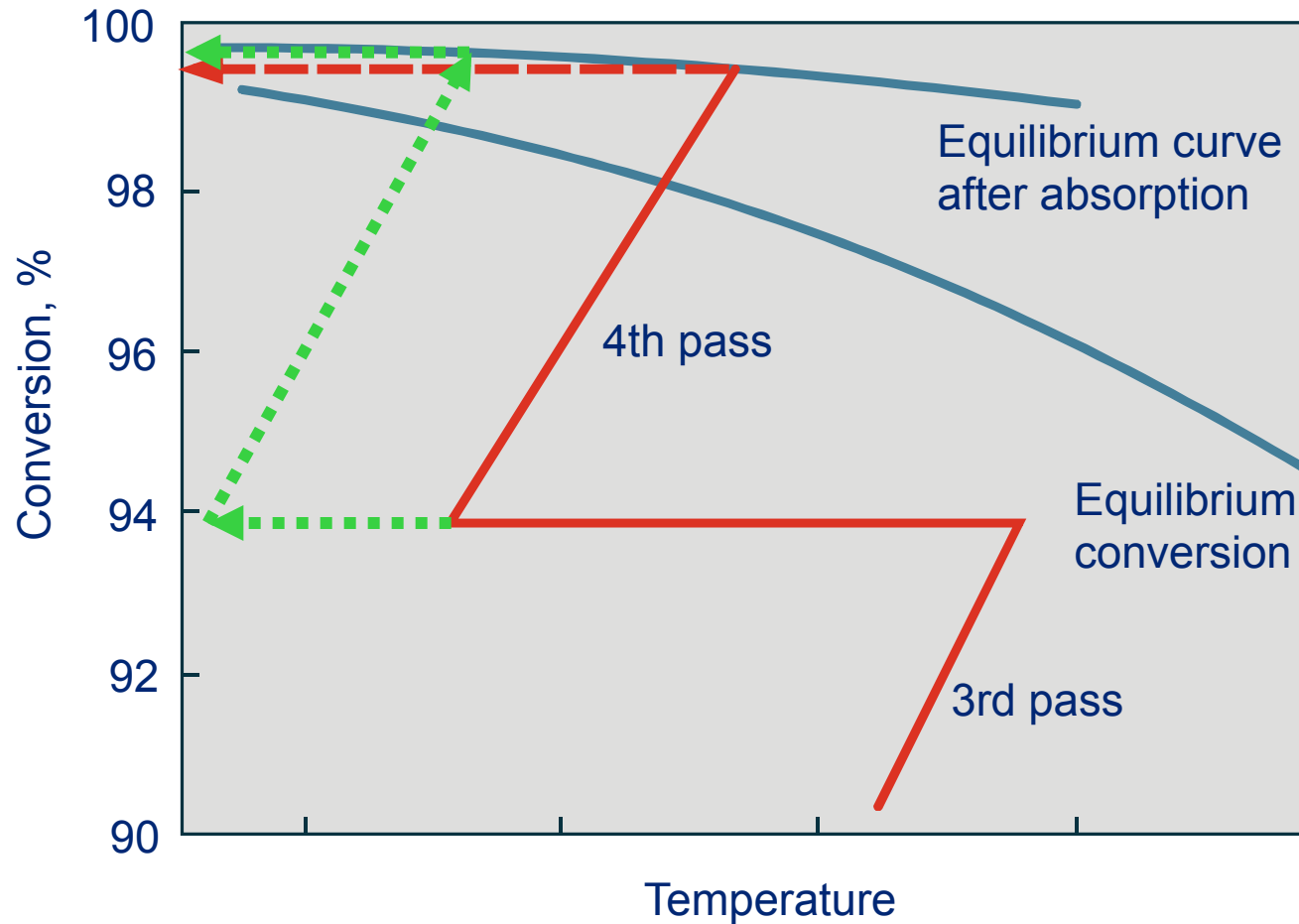
- Introduced in 1996 – 110+ references
- Designed specifically for double absorption plants in the bed after the intermediate absorption tower.
- Much higher activity at low temperatures
- Reduces emissions up to 50%

Activity in low SO₂ & O₂ gasses



Sulphur dioxide oxidation

Effect of interstage absorption on equilibrium conversion



VK 701 LEAP5™

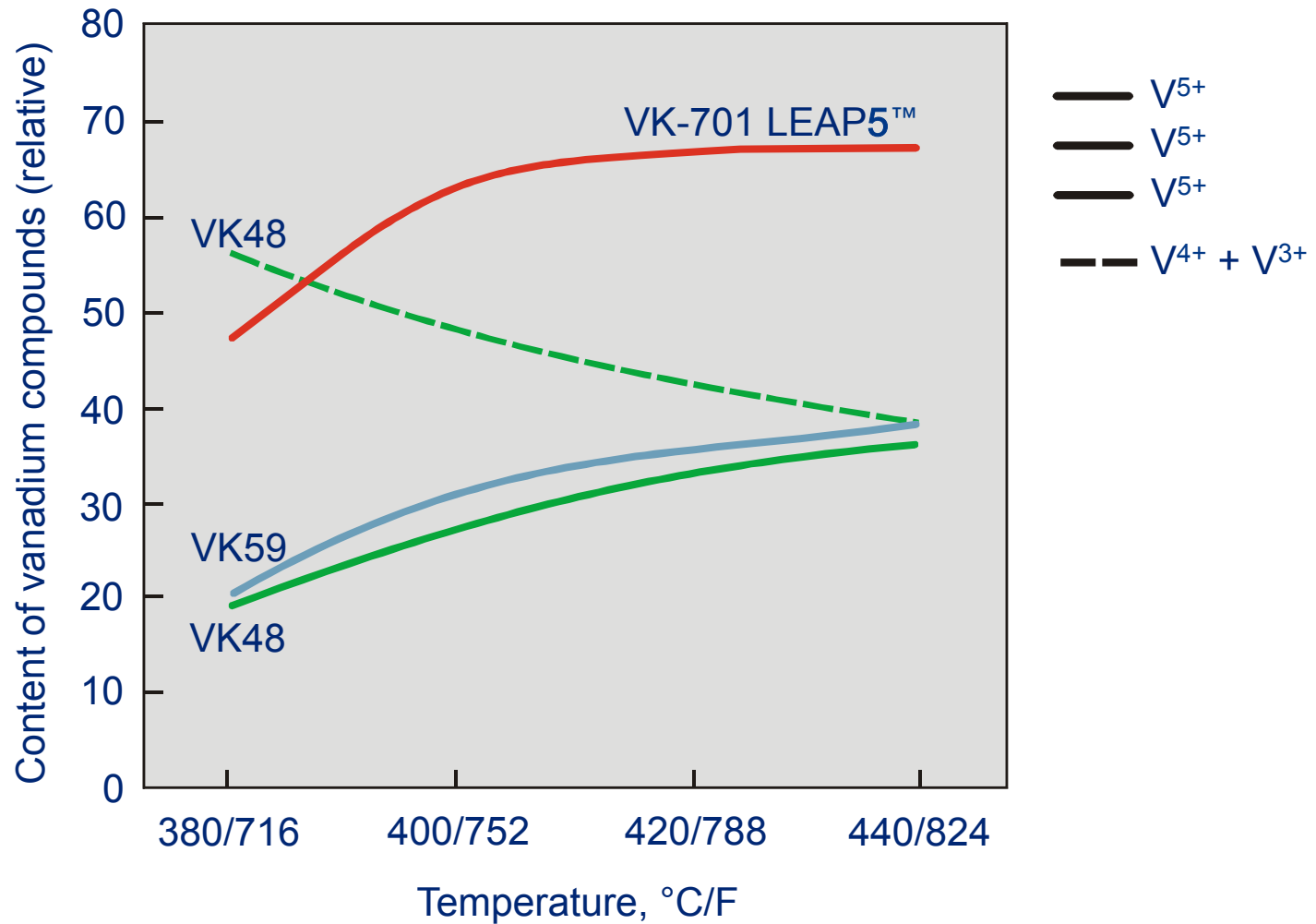


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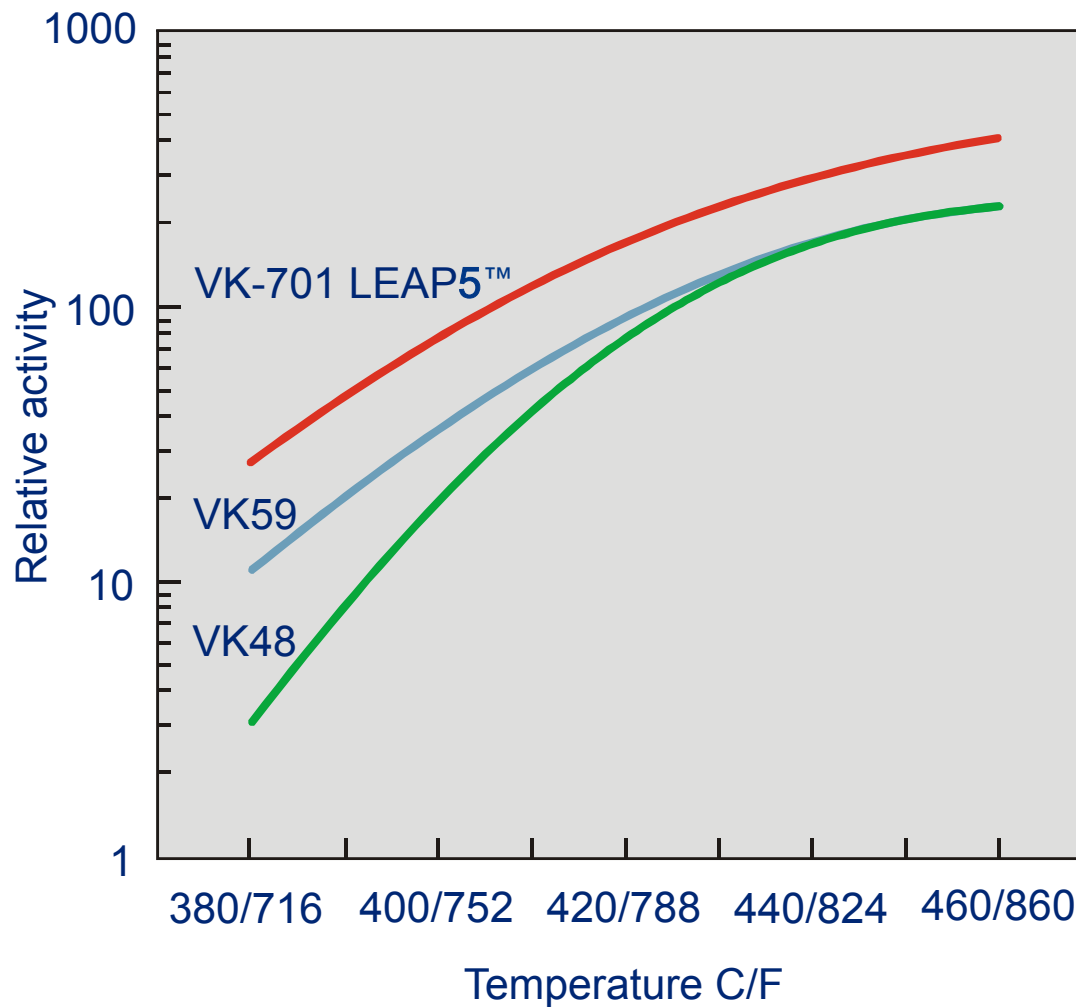
VK-701 LEAP5™

- Changed intrinsic morphology and surface properties of the carrier
- Optimized active phase for high SO₃ concentration
- Developed a new unique production technology
- Reduces emissions up to 40%

Vanadium oxidation states in 3rd bed catalysts

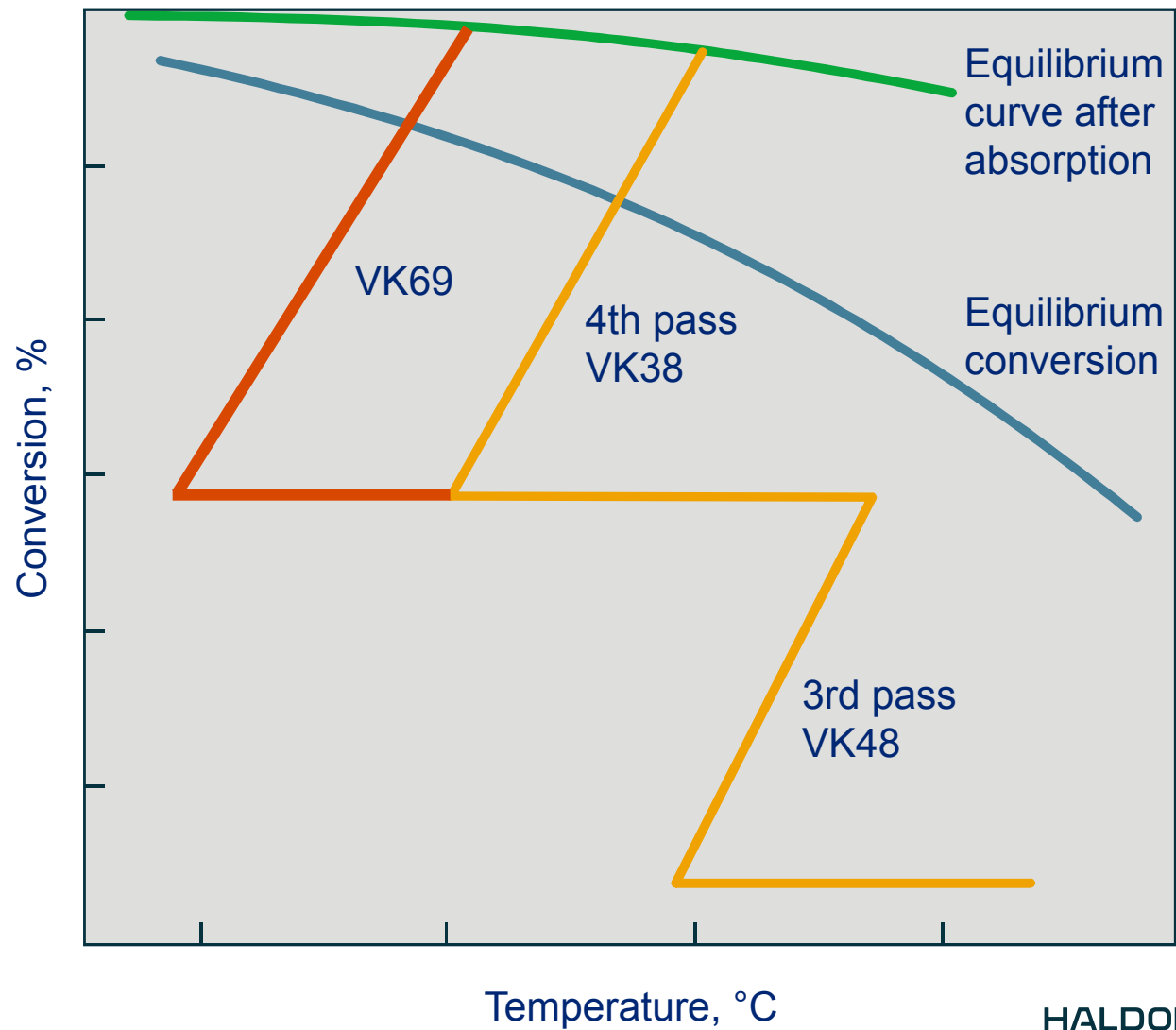


Activity of VK-701 LEAP5™ in SO₃ rich gas



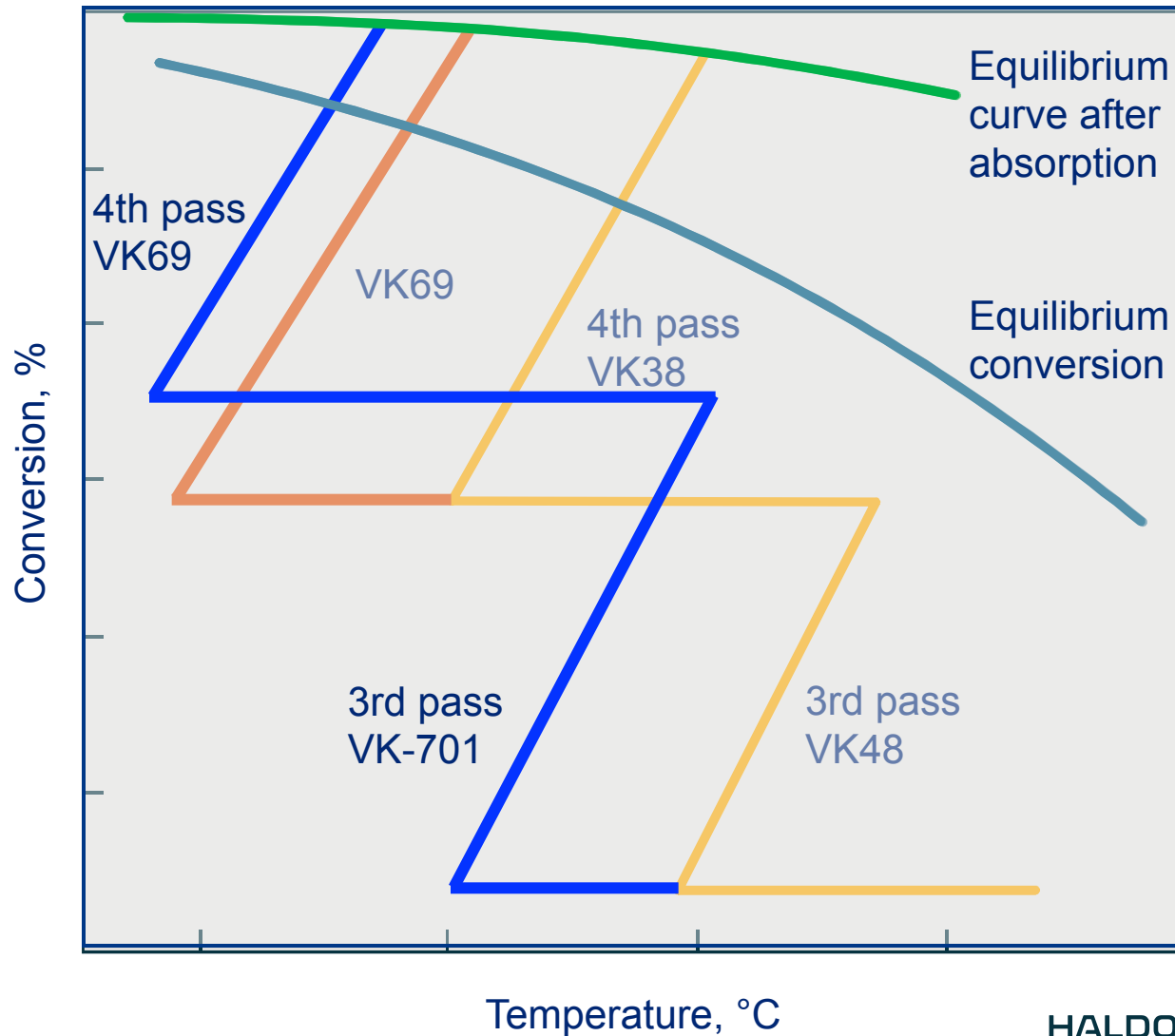
Equilibrium conversion

- VK69 application



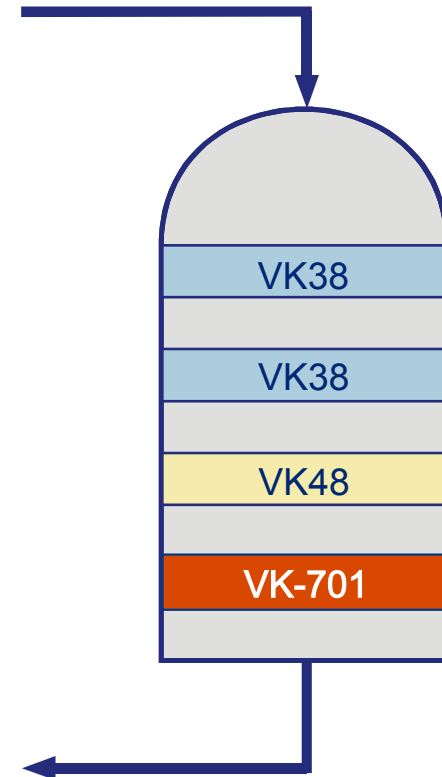
Equilibrium conversion

- VK69 & VK701 LEAP5™ applications



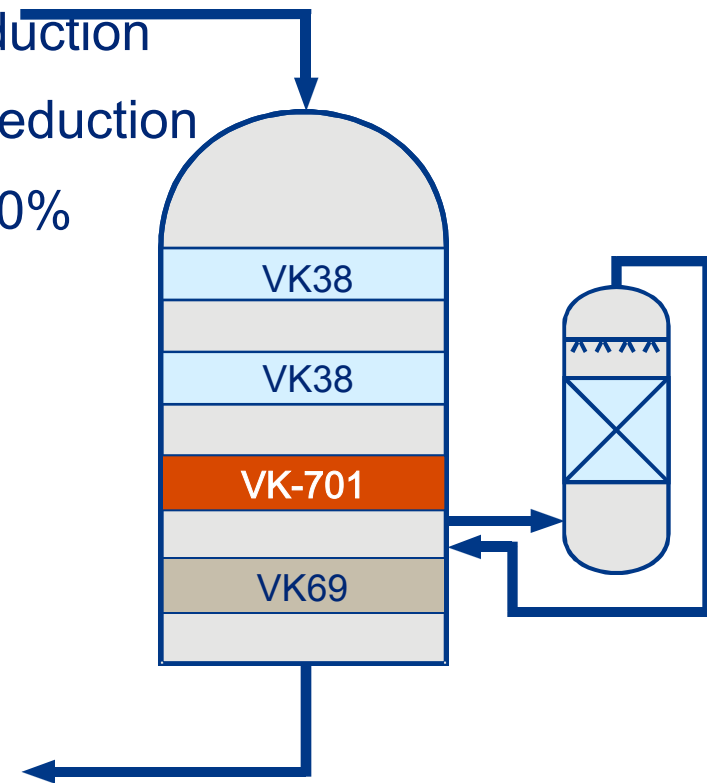
Single Absorption Example

Layout	: 4-pass single-absorption
SO ₂ source	: Metallurgical off-gas
Catalysts in beds 1/2/3	: VK38 / VK38 / VK48
Catalyst in bed 4	: VK-701
Emissions reduction	: Approximately 40%



Double Absorption Example

Layout	: 3+1 double-absorption plant
SO ₂ source	: S-burning in air
Catalysts in beds 1/2	: VK38 / VK38
Catalyst in bed 4	: VK69 ~ 50% reduction
Catalyst in bed 3	: VK-701 ~ 40% reduction
Net reduction	: approximately 70%



References



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VK-701 LEAP5™ References

- Single absorption sulfur burning
- Single absorption smelter off-gas
- Double absorption sulfur burning

Lessons Learned



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Lessons Learned

- No heat exchanger modifications have been required.
- Optimum inlet temperature only 25 to 30F lower for existing plants.
- VK-701 was “very hard” after being in service for a turnaround cycle.

Conclusion

- Demand to decrease emissions will continue
- Use of optimized catalysts will result in lower emissions
- No heat exchanger modifications required to date.

Conclusion (cont)

- Emission reductions through catalysis
 - Does not require additional energy consumption
 - Does not create any waste - the only byproduct is additional sulfuric acid production
 - Has the potential for additional energy recovery
 - Cost effective

Conclusion (cont)

- Potential emissions reduction
 - Single absorption
 - VK701 – approximately 40%
 - Double absorption
 - VK69 – approximately 50%
 - VK69 and VK701 – approximately 70%

Thank you

Any questions?